

**Supplementary information for**

**Flow-through Microfluidic Photoionization Detectors for**

**Rapid and Highly Sensitive Vapor Detection**

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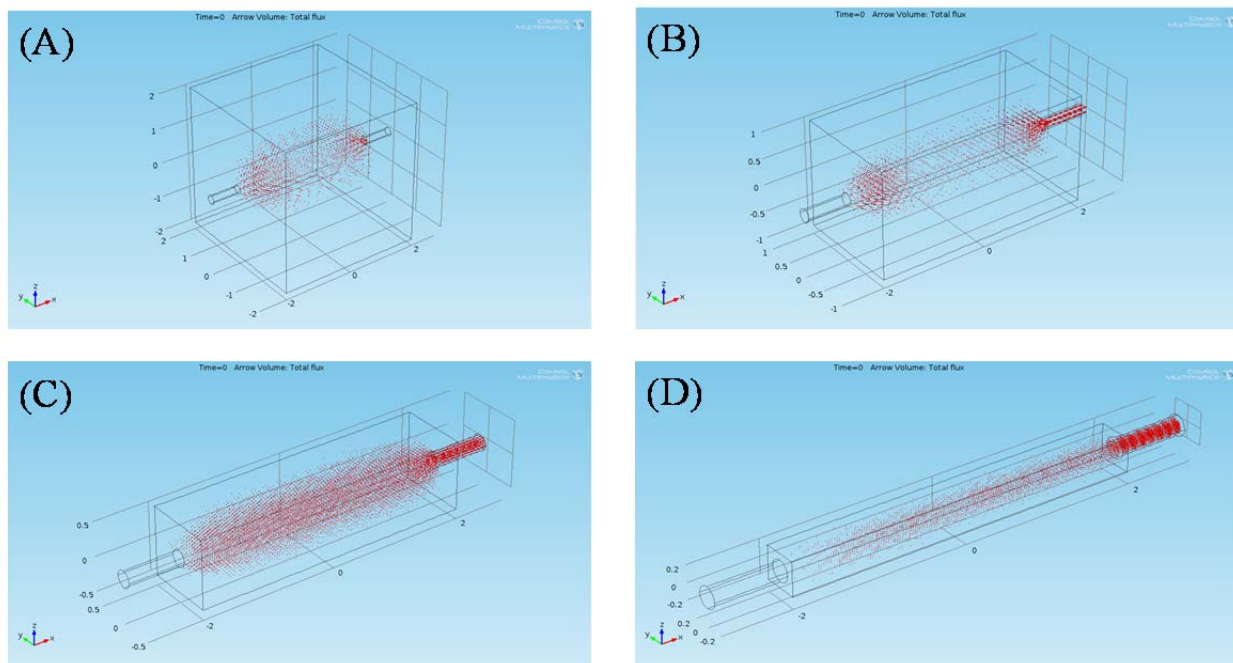


Fig. S1. COMSOL simulation of analyte (toluene) concentration (flux) magnitude for various sizes of chambers. (A)  $4 \times 4 \times 4 \text{ mm}^3 = 64 \text{ }\mu\text{L}$ . (B)  $2 \times 2 \times 4 \text{ mm}^3 = 16 \text{ }\mu\text{L}$ . (C)  $1 \times 1 \times 4 \text{ mm}^3 = 4 \text{ }\mu\text{L}$ . (D)  $0.4 \times 0.4 \times 4 \text{ mm}^3 = 0.64 \text{ }\mu\text{L}$ . The inlet and outlet have a diameter of 0.25 mm. Initially, the chamber is filled homogeneously with toluene. Purging gas, helium, is flowed in at  $t=0$  at a flow rate of 5 mL/min to purge the chamber. The dead volume, defined as the region inside the chamber having a helium flow velocity less than 10% of the maximal velocity, is estimated to be 35.7  $\mu\text{L}$ , 6.55  $\mu\text{L}$ , 1.57  $\mu\text{L}$ , and 0.24  $\mu\text{L}$ , respectively, for (A)-(D).

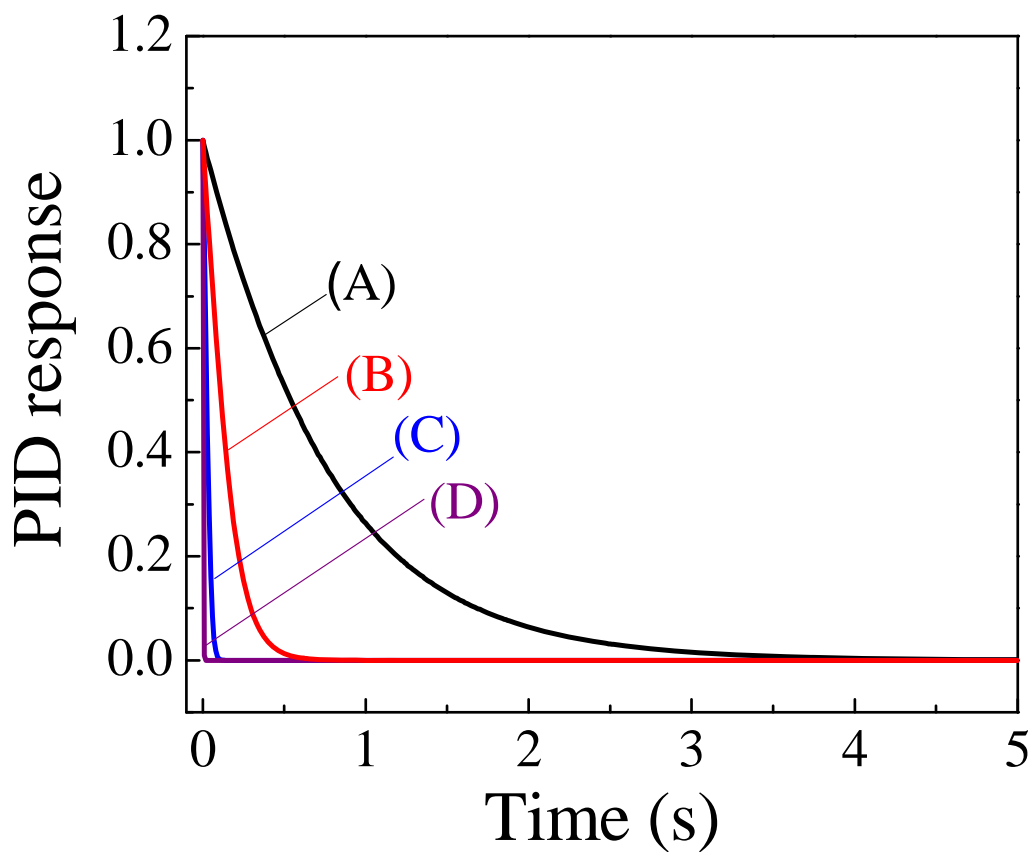


Fig. S2. Normalized PID response (proportional to the volume-averaged toluene concentration inside the chamber) as a function of purging time for various chamber sizes. Curves (A)-(D) correspond to Figs. S1(A)-(D). The fall time (*i.e.*, the time from the peak to 10% of the peak) is 1.69 s, 0.3 s, 0.049s, and 0.0035 s, respectively, for Curves (A)-(D).

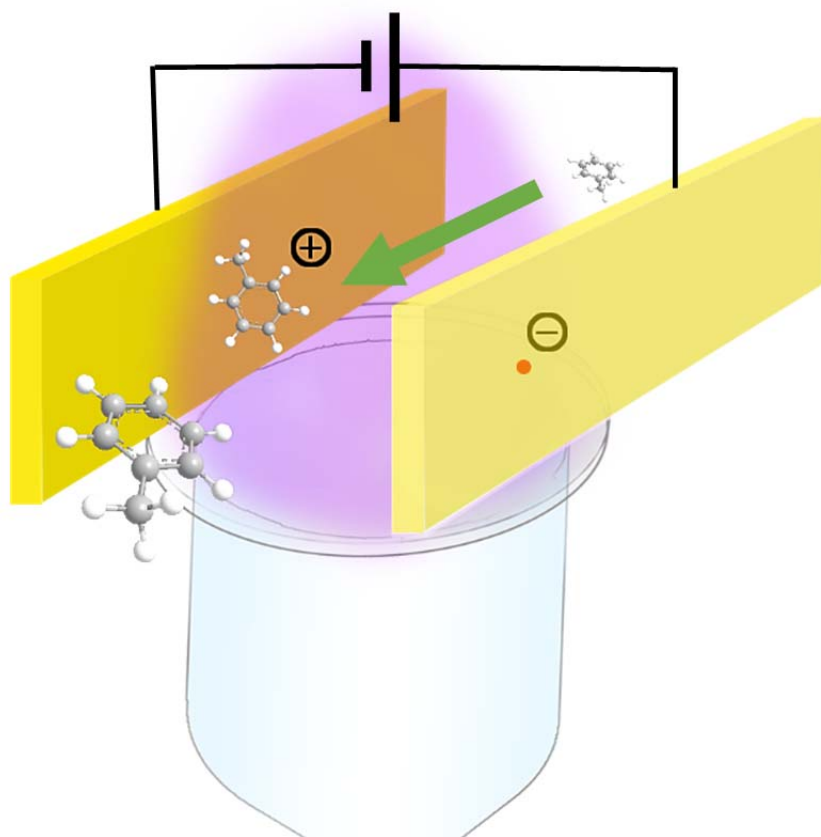


Fig. S3. PID working principle.

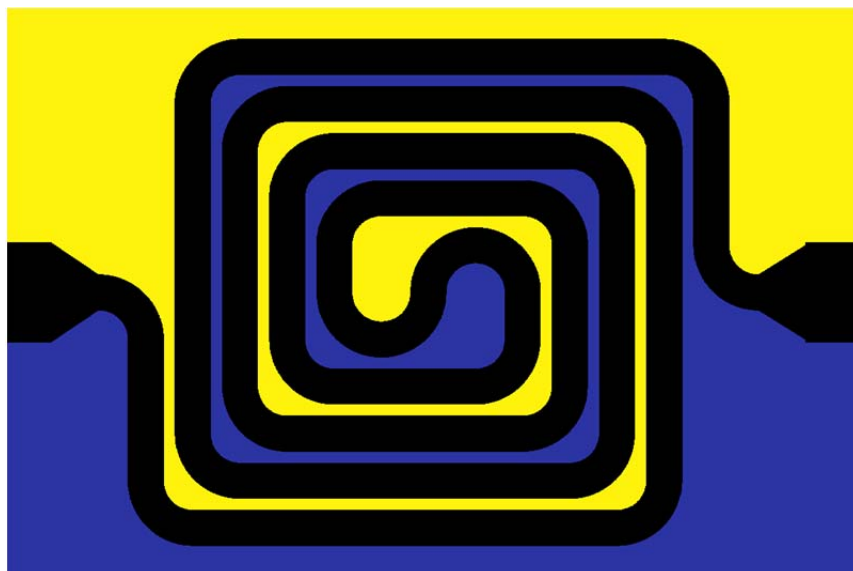


Fig. S4. Electrode layout for the microfluidic PID. Blue and yellow: electrodes. Black: microfluidic channel.

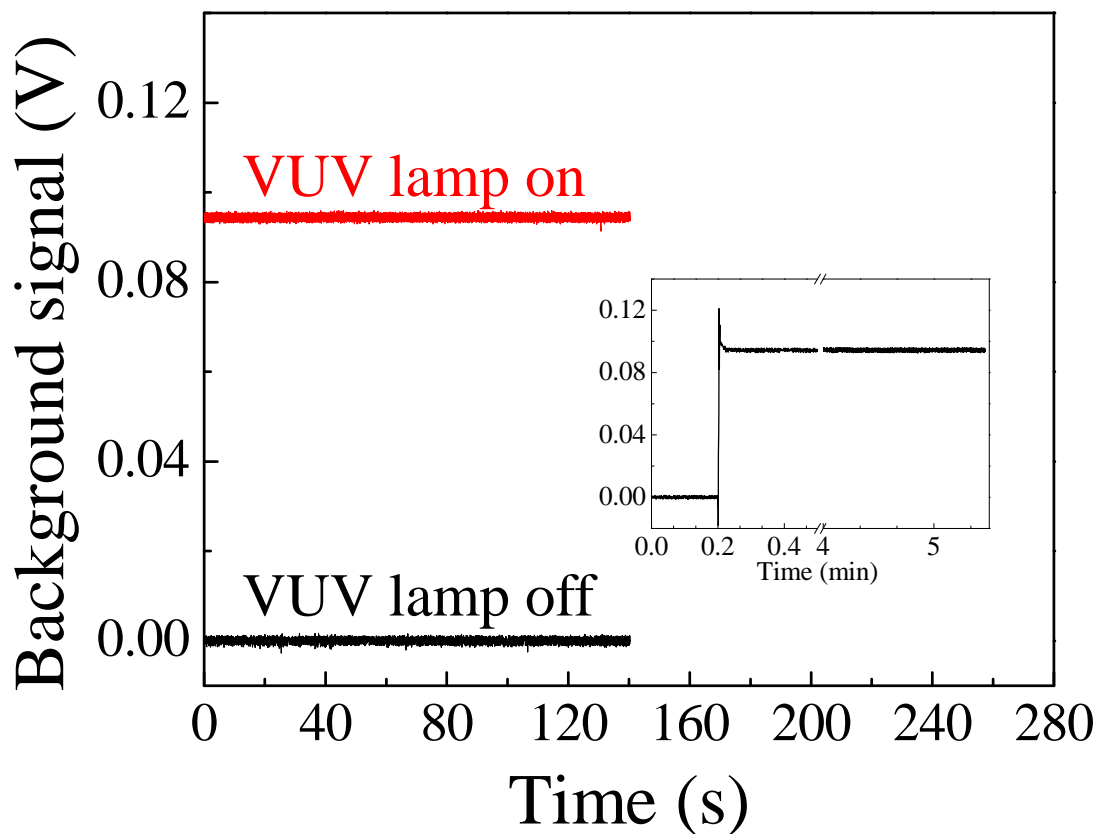


Fig. S5. Voltage signal increases by 94.3 mV when the VUV lamp is turned on, which corresponds to a 94.3 pA current change prior to the amplifier. The standard deviation of the noise is 0.68 mV. Amplification = 10X. Internal resistance of the amplifier = 100 M $\Omega$  + 25 pF. During the measurement, helium was flowed through the microfluidic PID at a flow rate of 2 mL/min. Inset shows long-term stability up to 5.5 minutes.

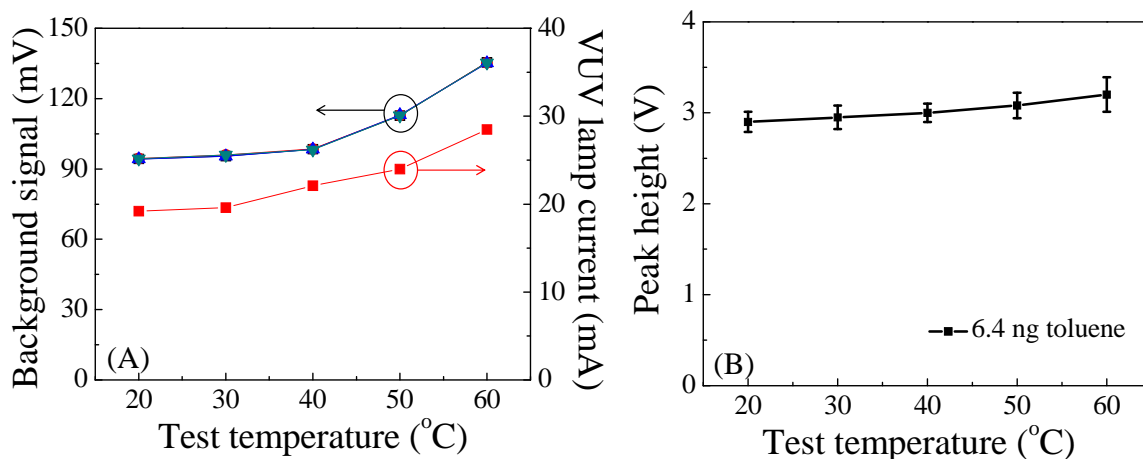


Fig. S6. Microfluidic PID temperature stability tests. (A) Left Y-axis: Baseline signal as a function of device temperature. The noise level remains the same as for 20 °C (Fig. S5). Right Y-axis: Current of the VUV lamp drive circuit. (B) PID sensitivity to analyte as a function of temperature. The corresponding baselines are removed. Error bars are calculated based on 4 measurements. During the test, the entire device was placed inside a GC oven. 60 °C is the maximal operating temperature specified by the VUV lamp manufacturer.

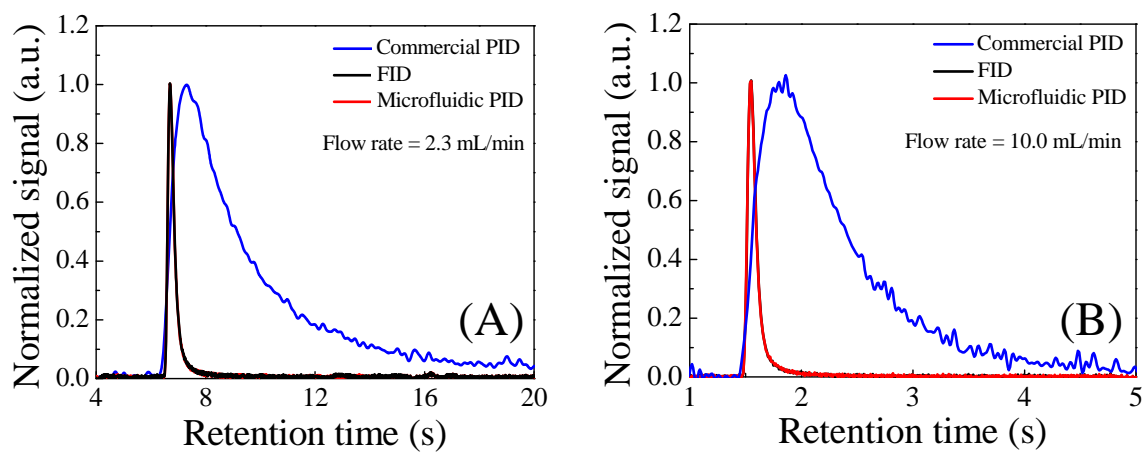


Fig. S7. Normalized toluene peak obtained with commercial PID, FID, and microfluidic PID at the flow rate of 2.3 mL/min and 10 mL/min, respectively. The corresponding magnified curves for FID and microfluidic PID are plotted in Fig. 3.



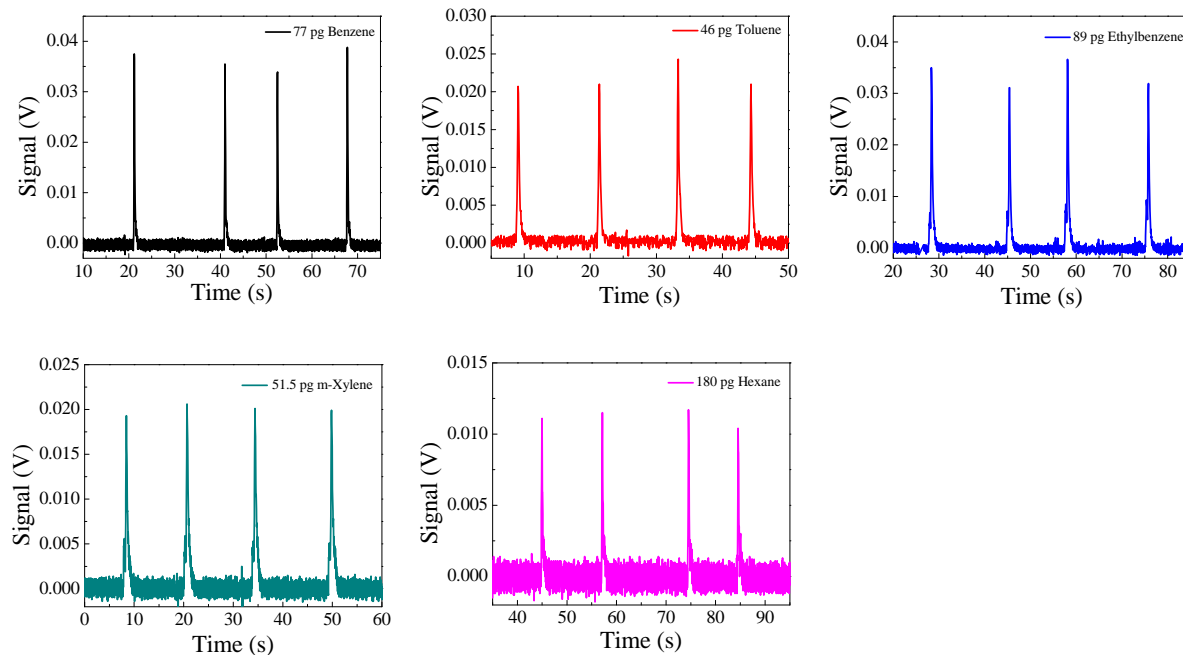


Fig. S8. Microfluidic PID signals for repeatable measurements (4 times) of various analytes that were injected into the GC-PID system.

Table S1. Comparison among micro-PID, plasma-based PID, and microfluidic PID. Red highlights denote meritorious or preferred features.

|  | <b>μPID<br/>(reported in Ref. 1)</b>                                       | <b>Plasma-based PID<br/>(Reported in Ref. 2)</b>      | <b>Microfluidic PID<br/>(this work)</b>  |
|--|--|---|--|
| <b>Design concept</b>                              | Traditional design concept, but with a smaller chamber volume              | New concept with micro-discharge integrated on a chip | New microfluidic flow-through concept. Low chamber volume and dead volume. Large electrode area. Low voltage |
| <b>Ionization source</b>                           | UV lamp on top of the ionization chamber (10.6 eV)                         | UV + Plasma (~20 eV)                                  | UV lamp on top of the ionization channel (10.6 eV)   |
| <b>Discharge electrode degradation</b>             | No   | No (according to 12-hr continuous testing)            | No   |
| <b>Source/analyte contact</b>                      | Non-contact  | Contact   | Non-contact  |
| <b>External helium flow required</b>               | No   | Yes   | No   |
| <b>High voltage required</b>                       | Yes (150 V)  | Yes (>500 V)  | No (6 V)   |
| <b>Power consumption</b>                           | ~100 mW (mainly for VUV operation)   | 1.4 mW  | ~100 mW (mainly for VUV operation)   |
| <b>Detection limit</b>                             | 5 ppb for benzene, toluene, and styrene (no mass detection limit reported) | ~10 pg for n-Octane                                   | ~5 pg for benzene, toluene, ethylbenzene, and m-xylene   |
| <b>Chamber size</b>                                | 10 μL  | Not reported  | 1.3 μL   |
| <b>Response time (baseline to 90% of the peak)</b> | 30 ms for 30 mL/min flow rate  | 170 ms for 2.5 mL/min flow rate                       | 160 ms and 55 ms for 2.3 and 10 mL/min flow rate, respectively   |
| <b>Dynamic range</b>                               | 6 orders of magnitude  | Not reported  | 6 orders of magnitude  |
| <b>Linearity</b>                                   | Non-linear (square root)   | Not reported  | Linear   |
| <b>Integration with μGC systems</b>                | Acceptable   | Best  | Better   |
| <b>Temperature stability</b>                       | Tested (25 °C)   | Tested (40-100 °C)                                    | Tested (20-60 °C)  |
| <b>Flow rate stability</b>                         | Tested (1.5 and 30 mL/min)   | Tested (0.9– 2.7 mL/min)                              | Tested (2.3 – 10 mL/min)   |
| <b>Undesirable negative peaks in signal</b>        | Not present  | Present (reasons unknown)                             | Not present  |

Table S2. Linear curve fit parameters used in Fig. 5(A).

|                                     | <b>Benzene</b>    | <b>Toluene</b>    | <b>Ethylbenzene</b> | <b>m-Xylene</b>   | <b>Hexane</b>     |
|-------------------------------------|-------------------|-------------------|---------------------|-------------------|-------------------|
| <b>Slope<br/>(Vs per ng)</b>        | 0.10452           | 0.07719           | 0.04908             | 0.05752           | 0.01274           |
| <b>R<sup>2</sup></b>                | 0.9856            | 0.96264           | 0.96136             | 0.9734            | 0.98119           |
| <b>Molecular weight<br/>(g/mol)</b> | 78                | 92                | 106                 | 106               | 86                |
| <b>Slope<br/>(Vs per mol)</b>       | $7.8 \times 10^9$ | $7.1 \times 10^9$ | $5.2 \times 10^9$   | $6.1 \times 10^9$ | $1.1 \times 10^9$ |

### Transit time calculation

Assuming a uniform electric field, the time for ions to move from one electrode to another,  $t$ , is given by

$$t = \sqrt{\frac{2m}{eV}} \cdot L, \quad (\text{S1})$$

where  $m$  and  $e$  are the mass and charge of an ion, respectively.  $L$  is the distance between the two electrodes.  $V$  is the applied voltage.

## References

1. J. Sun, F. Guan, D. Cui, X. Chen, L. Zhang, and J. Chen, "An improved photoionization detector with a micro gas chromatography column for portable rapid gas chromatography system," *Sens. Actuators B* **188**, 513– 518 (2013).
2. M. Akbar, H. Shakeel, and M. Agah, "GC-on-Chip: Integrated Column and Photo Ionization Detector," *Lab Chip* **15**, 1748-1758 (2015).